



Machakos University

(University Examinations 2021/2022)

SCHOOL OF ENGINEERING

DEPARTMENT OF MECHANICAL AND MANUFACTURING ENGINEERING

EEE 201: THERMODYNAMICS

Time: 2 Hours

Instructions:

- This paper consists of FIVE questions
- Attempt Question One and any other TWO questions
- Maximum marks for each part of a question are as indicated.

Provided

- Molar (Universal) gas constant $R_0 = 8314.5 \text{ J/kmolK}$
- Thermodynamic and Transport Properties of Fluids, SI Units - by Rogers G.F.C and Mayhew Y.R

Question 1(30 marks)

a) Briefly define the following terms as used in thermodynamics:

- System
- Surrounding
- Universe
- Heat engine
- Cycle

[5 marks]

b) oxygen has a molecular weight of 32 and a specific heat at constant volume = 650 J/kg K. Determine:

- the ratio of the specific heats
- the change in internal energy and enthalpy if 2.2 kg of the Oxygen gas is heated from 25 °C to 135 °C.

Take universal gas constant to be $R_o = 8314.3 \text{ kJ/kol K}$

[5 marks]

c) A boiler is supplied with feed water at 12 kPa as saturated liquid and delivers steam at 3 bar and 400 °C. If the furnace of this boiler is oil fired, the calorific value of the oil being 44 MJ/kg; determine the efficiency of the combustion when 38000 kg of oil is supplied to generate 27000 kg of steam.

[5 marks]

d) With the aid of a T-s diagram, describe the processes involved in the thermodynamic Carnot cycle.

[5 marks]

e) A power utility company desires to use the hot ground water from a hot spring to power a heat engine. If the ground water is at 95 °C, estimate the maximum power output if a mass flow rate of 0.25 kg/s is possible. The atmosphere is at 20 °C.

[5 marks]

f) Briefly explain the following terms as applied in heat transfer applications:

- Conduction
- Forced convection
- Black body
- Reflectivity
- Absorptivity

[5 marks]

Question 2 (20 marks)

a) Define the following thermodynamic processes:

- Isobaric process
- Adiabatic process
- Isothermal process

[3 marks]

b) Show that the heat transferred, Q , during a polytropic process is given by;

$$Q = \left(\frac{n - \gamma}{\gamma - 1} \right) W_{\text{poly}}$$

Where, W_{poly} – work done during a polytropic process, $p v^n = \text{constant}$

n – polytropics exponent

γ - adiabatic index

[8 marks]

c) A mass 3 kg of air is compressed through a polytropic process from an initial state of pressure 101 kPa and temperature 27 °C to a final state of pressure and temperature of 350 kPa and 325 °C, respectively. Determine the polytropic exponent and the heat transferred during the compression process. Take $R = 0.287$ kJ/kg K and $\gamma = 1.4$.

[9 marks]

Question 3 (20 marks)

a) Give a brief description to the following types of thermodynamic systems:

- Open system
- Closed system
- Isolated system

[3 marks]

b) Show that the change of entropy during a constant pressure process can be obtained using the expression;

$$S_2 - S_1 = m c_p \ln \frac{v_2}{v_1}$$

Where; m – mass of the gas

v_1 – Initial specific volume of the gas

T_1 – Initial temperature of the gas

v_2, T_2 – corresponding values for the final condition

[7 marks]

- c) 1.5 kg of air is heated at constant pressure of 4 bar to 450 °C. if the initial volume is 0.8 m³, determine;

- The initial temperature
- The final volume
- The heat exchanged
- The change in entropy

Take $R = 287 \text{ J/kg K}$ and $\gamma = 1.4$.

[10 marks]

Question 4 (20 marks)

- a) Differentiate between extensive and intensive thermodynamic properties. [2 marks]

- b) Explain the following terms as used in steam utilizing systems;

- Wet steam
- Dryness fraction
- Degree of superheat

[3 marks]

- c) Steam enters the turbine in a power plant at 40 bar and 400 °C and is condensed at a pressure of 1 bar till it exits the condenser as saturated liquid.

- i. Assuming the plant can be treated as an ideal Rankine cycle; determine the power produced per kg of steam and the efficiency of the cycle.
- ii. Determine the efficiency of a Carnot cycle operated between these two temperatures.

[15 marks]

Question 5 (20 marks)

- a) Show that the critical radius of insulation of a cylindrical pipe is given by the expression:

$$r_{cr} = \frac{k}{h}$$

Where; k – thermal conductivity of insulation material

h – convection heat transfer coefficient

[5 marks]

- b) A 4-mm-diameter and 8-m-long electric wire is tightly wrapped with a 3-mm-thick plastic cover whose thermal conductivity is $k = 0.15 \text{ W/m.K}$. Electrical measurements indicate that a current of 10 A passes through the wire and there is a voltage drop of 6 V along the wire. If the insulated wire is exposed to the ambient which is at 28 °C with a heat transfer coefficient of $h = 12 \text{ W/m}^2\text{.K}$, determine under steady operating conditions:

- i. The interface temperature of the wire and the plastic cover
- ii. What will be the interface temperature of the wire and the plastic cover if the critical radius of insulation is used?

[7 marks]

- c) Stating the assumptions made, show that the overall heat transfer in a parallel-flow heat exchanger is given by the expression;

$$q = U A \Delta T_{lm}$$

Where; U = overall heat transfer coefficient

A = surface area across which heat transfer takes place

ΔT_{lm} = Logarithmic mean temperature difference. [8 marks]