



MACHAKOS UNIVERSITY

University Examinations for 2020/2021 Academic Year

SCHOOL OF ENGINEERING AND TECHNOLOGY

DEPARTMENT OF MECHANICAL AND MANUFACTURING ENGINEERING

FOURTH YEAR SPECIAL/SUPPLEMENTARY EXAMINATION FOR

BACHELOR OF SCIENCE (MECHANICAL ENGINEERING)

EMM 404: ENGINEERING THERMODYNAMICS III

DATE: 22/3/2021

TIME: 2.00-4.00 PM

INSTRUCTIONS

Answer Question One and Any Other Two Questions

QUESTION ONE (30 MARKS) (COMPULSORY)

- a) Briefly define the following terms as applied in reciprocating compressors:
- i. Compression ratio
 - ii. Swept volume
 - iii. Clearance volume
 - iv. Compressor capacity
 - v. Free air delivery (5 marks)
- b) Represent the processes making up the diesel cycle on P-v and T-s diagrams. (5 marks)
- c) An internal combustion four-cylinder engine has a bore of 7.2 cm, a stroke of 7.8 cm, and a clearance length of 1 cm, determine the compression ratio and engine displacement (5 marks)
- d) The percentage composition of a sample of coal is found by mass analysis as: C 90%, H₂ 4%, O₂ 1.5%, N₂ 1.3%, S 0.8% and the remaining is ash.
Calculate the theoretical mass of air for complete combustion of 1 kg of coal. Take air to consist of 23% O₂ by mass. (5 marks)

- e) State FIVE assumptions made in the analysis of the thermal efficiency of an ideal gas turbine plant. (5 marks)
- f) Starting with the fundamental thermodynamic relations for the First law and entropy with specific reference to internal energy, obtain the Maxwell's relation.

$$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial P}{\partial S}\right)_V$$

Where; T – temperature

V – volume

P – pressure

S – entropy

(5 marks)

QUESTION TWO (20 MARKS)

- a) Deduce the expression for the thermal efficiency of the air standard Diesel cycle in terms of the compression and cut-off ratios. (10 marks)
- b) An air- standard Diesel cycle has a compression ratio of 16 and a cut-off ratio of 2. At the beginning of the compression process, air is at 95 kPa and 27 °C. Determine:
- The temperature after the heat-addition process
 - The thermal efficiency
 - The mean effective pressure

(10 marks)

Take $\gamma = 1.4$

QUESTION THREE (20 MARKS)

A single acting, single stage air compressor is required to deal with 8 m³ of air per minute measured at 1.013 bar and 25°C. The conditions at the beginning of compression are 0.96 bar and 28 °C and the delivery pressure is 5 bar. The compressor clearance volume is 6% of the swept volume and the operating speed is 360 rev/min. The stroke is 1.2 times the cylinder diameter. The index of compression and expansion is 1.32.

Calculate;

- a) the power required
- b) the cylinder diameter
- c) the isothermal efficiency of the compressor.

Take $c_p = 1.005$ kJ/kg K, and $\gamma = 1.4$

QUESTION FOUR (20 MARKS)

A gas turbine unit takes in air at 20°C and 100 kPa and operates with a pressure ratio of 18:1. The maximum combustor temperature is 2800 K at a mass flow rate of 8.8 kg/s. Assuming ideal compression and expansion processes, determine;

- a) The compressor power input
- b) The rate of heat addition
- c) The net power output
- d) The thermal efficiency for the cycle
- e) The work ratio

Take: $c_p = 1.005 \text{ kJ/kg K}$, and $\gamma = 1.4$

QUESTION FIVE (20 MARKS)

- a) The ultimate analysis of a sample of petrol was 85% C and 15% H. The analysis of the dry products gave 12% CO_2 and some O_2 . Determine;
 - i. The air/fuel ratio supplied to the engine
 - ii. The mixture strength.
- b) Derive the Clausius-Clapeyron equation for phase change.
- c) With the help of the Maxwell's relations and expression for Gibb's free energy, show that for an ideal gas the enthalpy (H) is independent of pressure (P).

Important Relations

Functions:

$$A = U - TS$$

$$G = H - TS$$

Differentials

$$dU = TdS - PdV$$

$$dH = TdS + VdP$$

$$dA = -PdV - SdT$$

$$dG = VdP - SdT$$

Maxwell's Relations

$$\left(\frac{\partial T}{\partial V} \right)_S = - \left(\frac{\partial P}{\partial S} \right)_V$$

$$\left(\frac{\partial T}{\partial P} \right)_S = \left(\frac{\partial V}{\partial S} \right)_P$$

$$\left(\frac{\partial P}{\partial T} \right)_V = \left(\frac{\partial S}{\partial V} \right)_T$$

$$\left(\frac{\partial V}{\partial T} \right)_P = - \left(\frac{\partial S}{\partial P} \right)_T$$